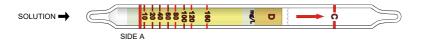
WATER CONTENT IN SOLVENT



1. PERFORMANCE

1) Sampling method : Suction method

2) Measuring range : 10-160 mg/L 50-400 mg/L

(draw sample up to C) (draw sample up to D)

3) Sampling time : 10 seconds4) Sample volume : Approx. 1 mL5) Detectable limit : 5 mg/L6) Shelf life : 2 years7) Operating temperature $: 10 \sim 30 ^{\circ}\text{C}$

8) Reading : Direct reading from the scale (draw sample up to C)

9) Colour change : Yellow → Blueish purple

2. RELATIVE STANDARD DEVIATION

RSD-low: 10% RSD-mid.: 10% RSD-high: 10%

3. CHEMICAL REACTION

By reacting with Magnesium perchlorate, double salt is produced. $H_2O + Mg(CIO_4)_2 \rightarrow Mg(CIO_4)_2 \cdot H_2O$

4. CALIBRATION OF THE TUBE

KARL FISCHER TITRATION

5. INTERFERENCE AND CROSS SENSITIVITY

Compatible solvents	Benzene, Toluene, Xylene, Styrene, Hexane, Trichloroethylene Tetrachloroethylene, Carbon tetrachloride, Fureon, Gasoline Kerosine, Naphtha, JP-4
Incompatible solvents	Alcohols, Esters, Nitrobenzene, Chloroform, 1,1,1-Trichloroethane 1,2-Dichloroethane, Ethane tetrachloride, Dioxane, Tetrahydrofuran Ethyl cellosolve

6. SAMPLING METHOD

(Suction method)

- 1) Flaw both ends of a fresh detector tube with an ampule cutter.
- 2) Break the tube end (B), squeeze the rubber bulb (optional), insert the tube end (B) into the rubber bulb.
- 3) Break the tube end (A) and immerse the tube end (A) into sample solution immediately.
- 4) Put the thumb off the rubber bulb, and the sample solution is rose through reagents.
- 5) When the sample solution rises up to (C) of the tube, remove the tube from the rubber bulb and from the sample solution to take the reading.
- 6) When the concentration is over 160mg/L, repeat 1) through 4).
- 7) When the sample solution rises up to (D), remove the tube from the rubber bulb and from the sample solution.
- 8) The actual concentration can be obtained by using a conversion chart.

